

## Dimethyl(4'-pyridyl-2,2':6',2''-terpyridine- $\kappa^3N^1,N^1',N^1''$ )bis(thiocyanato- $\kappa N$ )-tin(IV)

Ezzatollah Najafi,<sup>a</sup> Mostafa M. Amini<sup>a</sup> and Seik Weng Ng<sup>b\*</sup>

<sup>a</sup>Department of Chemistry, General Campus, Shahid Beheshti University, Tehran 1983963113, Iran, and <sup>b</sup>Department of Chemistry, University of Malaya, 50603

Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

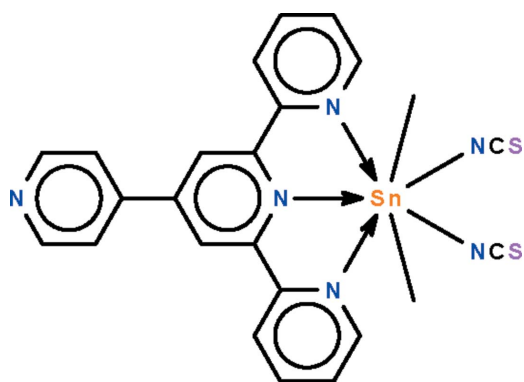
Received 9 January 2011; accepted 10 January 2011

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.024;  $wR$  factor = 0.060; data-to-parameter ratio = 17.2.

The Sn atom in the title compound,  $[\text{Sn}(\text{CH}_3)_2(\text{NCS})_2(\text{C}_{20}\text{H}_{14}\text{N}_4)]$ , is  $N,N',N''$ -chelated by the terpyridine part of the  $N$ -heterocycle. The Sn atom exists in a *trans*- $\text{C}_2\text{SnN}_5$  pentagonal-bipyramidal geometry [ $\text{C}-\text{Sn}-\text{C} = 173.66(8)^\circ$ ] with the methyl groups in axial and the N atoms in equatorial positions.

### Related literature

For the dimethyltin dichloride-terpyridine adduct, see: Naik & Scheidt (1973).



### Experimental

#### Crystal data

$[\text{Sn}(\text{CH}_3)_2(\text{NCS})_2(\text{C}_{20}\text{H}_{14}\text{N}_4)]$

$M_r = 575.27$

Triclinic,  $P\bar{1}$

$a = 9.3269(3)$  Å

$b = 10.5017(3)$  Å

$c = 13.1503(4)$  Å

$\alpha = 66.814(3)^\circ$

$\beta = 87.665(3)^\circ$

$\gamma = 83.552(2)^\circ$

$V = 1176.52(6)$  Å<sup>3</sup>

$Z = 2$

Mo  $K\alpha$  radiation

$\mu = 1.29$  mm<sup>-1</sup>

$T = 100$  K

$0.20 \times 0.20 \times 0.10$  mm

#### Data collection

Agilent Technologies SuperNova diffractometer with an Atlas detector

Absorption correction: multi-scan (*CrysAlis PRO*; Agilent Technologies, 2010)

$T_{\min} = 0.783$ ,  $T_{\max} = 0.882$

15704 measured reflections

5168 independent reflections

4728 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.024$

$wR(F^2) = 0.060$

$S = 1.04$

5168 reflections

300 parameters

H-atom parameters constrained

$\Delta\rho_{\text{max}} = 0.51$  e Å<sup>-3</sup>

$\Delta\rho_{\text{min}} = -0.49$  e Å<sup>-3</sup>

Data collection: *CrysAlis PRO* (Agilent Technologies, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

We thank Shahid Beheshti University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5461).

### References

- Agilent Technologies (2010). *CrysAlis PRO*. Agilent Technologies, Yarnton, England.
- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
- Naik, D. N. & Scheidt, W. R. (1973). *Inorg. Chem.* **12**, 272–276.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

**supplementary materials**

*Acta Cryst.* (2011). E67, m211 [ doi:10.1107/S1600536811001309 ]

**Dimethyl(4'-pyridyl-2,2':6',2''-terpyridine- $\kappa^3N^1,N^1',N^1''$ )bis(thiocyanato- $\kappa N$ )tin(IV)**

**E. Najafi, M. M. Amini and S. W. Ng**

**Comment**

Diorganotin dihalides/pseudohalides form a number of adducts with terpyridine and its derivatives. The dimethyltin diisocyanate forms an adduct with terpyridine itself (Naik & Scheidt, 1973); the ligand chelates to the tin atom through its three N atoms. The 4'-pyridylterpyridine has a similar seven-coordinate structure (Scheme I, Fig. 1). It also features the chelated tin atom in a seven-coordinate geometry.

**Experimental**

Dimethyltin diisothiocyanate and 4'-pyridyl-2,2':6',2''-terpyridine (1 mmol) were loaded into a convection tube. The tube was filled with dry methanol and kept at 333 K. Colorless crystals were collected from the side arm after several days.

**Refinement**

H-atoms were placed in calculated positions [C—H 0.95 to 0.98 Å,  $U_{iso}(H)$  1.2 to 1.5 $U_{eq}(C)$ ] and were included in the refinement in the riding model approximation.

**Figures**

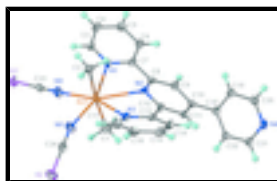


Fig. 1. Anisotropic displacement ellipsoid plot (Barbour, 2001) of diisothiocyanatodimethyl(4'-pyridylterpyridine)tin at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

**Dimethyl(4'-pyridyl-2,2':6',2''-terpyridine- $\kappa^3N^1,N^1',N^1''$ )bis(thiocyanato- $\kappa N$ )tin(IV)**

*Crystal data*

[Sn(CH<sub>3</sub>)<sub>2</sub>(NCS)<sub>2</sub>(C<sub>20</sub>H<sub>14</sub>N<sub>4</sub>)]

$M_r = 575.27$

Triclinic,  $P\bar{1}$

Hall symbol: -P 1

$a = 9.3269$  (3) Å

$b = 10.5017$  (3) Å

$c = 13.1503$  (4) Å

$\alpha = 66.814$  (3)°

$\beta = 87.665$  (3)°

$Z = 2$

$F(000) = 576$

$D_x = 1.624$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 10240 reflections

$\theta = 2.2$ – $29.4$ °

$\mu = 1.29$  mm<sup>-1</sup>

$T = 100$  K

Prism, colorless

# supplementary materials

---

$\gamma = 83.552 (2)^\circ$   
 $V = 1176.52 (6) \text{ \AA}^3$   
 $0.20 \times 0.20 \times 0.10 \text{ mm}$

## Data collection

Agilent Technologies SuperNova (Dual, Cu at zero) diffractometer with an Atlas detector 5168 independent reflections  
Radiation source: SuperNova (Mo) X-ray Source 4728 reflections with  $I > 2\sigma(I)$   
Mirror  $R_{\text{int}} = 0.028$   
Detector resolution:  $10.4041 \text{ pixels mm}^{-1}$   $\theta_{\text{max}} = 27.5^\circ$ ,  $\theta_{\text{min}} = 2.2^\circ$   
 $\omega$  scans  $h = -12 \rightarrow 11$   
Absorption correction: multi-scan (*CrysAlis PRO*; Agilent Technologies, 2010)  $k = -13 \rightarrow 13$   
 $T_{\text{min}} = 0.783$ ,  $T_{\text{max}} = 0.882$   $l = -17 \rightarrow 16$   
15704 measured reflections

## Refinement

Refinement on  $F^2$  Primary atom site location: structure-invariant direct methods  
Least-squares matrix: full Secondary atom site location: difference Fourier map  
 $R[F^2 > 2\sigma(F^2)] = 0.024$  Hydrogen site location: inferred from neighbouring sites  
 $wR(F^2) = 0.060$  H-atom parameters constrained  
 $S = 1.04$   $w = 1/[\sigma^2(F_o^2) + (0.027P)^2 + 0.2766P]$   
5168 reflections where  $P = (F_o^2 + 2F_c^2)/3$   
300 parameters  $(\Delta/\sigma)_{\text{max}} = 0.001$   
0 restraints  $\Delta\rho_{\text{max}} = 0.51 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.49 \text{ e \AA}^{-3}$

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{Å}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Sn1	0.670329 (15)	0.378196 (13)	0.343774 (11)	0.01570 (6)
S1	0.14249 (6)	0.55757 (6)	0.27483 (5)	0.02139 (12)
S2	0.54381 (7)	0.07411 (6)	0.72375 (5)	0.02716 (14)
N1	0.65080 (18)	0.52567 (17)	0.13746 (14)	0.0154 (4)
N2	0.88281 (18)	0.34503 (16)	0.22361 (14)	0.0137 (4)
N3	0.90522 (19)	0.26172 (17)	0.44304 (14)	0.0166 (4)
N4	1.26459 (19)	-0.03774 (18)	-0.05850 (15)	0.0184 (4)
N5	0.4323 (2)	0.45490 (19)	0.32114 (15)	0.0241 (4)
N6	0.5991 (2)	0.2807 (2)	0.51867 (15)	0.0256 (4)
C1	0.6332 (2)	0.1957 (2)	0.3225 (2)	0.0232 (5)
H1A	0.6343	0.2139	0.2434	0.035*
H1B	0.5391	0.1673	0.3536	0.035*
H1C	0.7091	0.1210	0.3606	0.035*
C2	0.7301 (2)	0.5579 (2)	0.35758 (18)	0.0201 (5)
H2A	0.7350	0.5426	0.4359	0.030*

H2B	0.6583	0.6369	0.3198	0.030*
H2C	0.8247	0.5779	0.3235	0.030*
C3	0.5515 (2)	0.6360 (2)	0.09447 (18)	0.0183 (4)
H3	0.4899	0.6621	0.1439	0.022*
C4	0.5342 (2)	0.7135 (2)	-0.01779 (18)	0.0200 (5)
H4	0.4632	0.7915	-0.0447	0.024*
C5	0.6229 (2)	0.6748 (2)	-0.08986 (17)	0.0185 (4)
H5	0.6121	0.7244	-0.1675	0.022*
C6	0.7274 (2)	0.5631 (2)	-0.04746 (17)	0.0166 (4)
H6	0.7900	0.5355	-0.0956	0.020*
C7	0.7396 (2)	0.4917 (2)	0.06618 (17)	0.0142 (4)
C8	0.8551 (2)	0.3767 (2)	0.11653 (17)	0.0138 (4)
C9	0.9287 (2)	0.3065 (2)	0.05728 (17)	0.0152 (4)
H9	0.9077	0.3327	-0.0188	0.018*
C10	1.0339 (2)	0.1968 (2)	0.11041 (17)	0.0146 (4)
C11	1.0643 (2)	0.1662 (2)	0.22092 (17)	0.0155 (4)
H11	1.1366	0.0935	0.2594	0.019*
C12	0.9879 (2)	0.2429 (2)	0.27423 (17)	0.0142 (4)
C13	1.0163 (2)	0.2199 (2)	0.39065 (17)	0.0158 (4)
C14	1.1508 (2)	0.1658 (2)	0.43977 (18)	0.0186 (5)
H14	1.2260	0.1346	0.4015	0.022*
C15	1.1733 (2)	0.1583 (2)	0.54597 (18)	0.0206 (5)
H15	1.2647	0.1238	0.5811	0.025*
C16	1.0606 (2)	0.2018 (2)	0.59894 (18)	0.0214 (5)
H16	1.0731	0.1974	0.6716	0.026*
C17	0.9282 (2)	0.2523 (2)	0.54579 (18)	0.0202 (5)
H17	0.8509	0.2813	0.5837	0.024*
C18	1.1117 (2)	0.1161 (2)	0.05122 (17)	0.0148 (4)
C19	1.1493 (2)	0.1789 (2)	-0.05975 (17)	0.0165 (4)
H19	1.1225	0.2753	-0.1005	0.020*
C20	1.2262 (2)	0.0990 (2)	-0.11010 (17)	0.0172 (4)
H20	1.2531	0.1439	-0.1853	0.021*
C21	1.2265 (2)	-0.0968 (2)	0.04727 (18)	0.0176 (4)
H21	1.2523	-0.1940	0.0853	0.021*
C22	1.1520 (2)	-0.0258 (2)	0.10523 (17)	0.0156 (4)
H22	1.1286	-0.0733	0.1809	0.019*
C23	0.3116 (2)	0.4974 (2)	0.30293 (17)	0.0173 (4)
C24	0.5756 (2)	0.1937 (2)	0.60383 (18)	0.0198 (5)

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Sn1	0.01512 (9)	0.01554 (8)	0.01612 (9)	-0.00096 (6)	0.00146 (6)	-0.00616 (6)
S1	0.0158 (3)	0.0253 (3)	0.0245 (3)	0.0018 (2)	0.0006 (2)	-0.0123 (2)
S2	0.0260 (3)	0.0268 (3)	0.0212 (3)	-0.0053 (2)	0.0007 (3)	-0.0008 (2)
N1	0.0135 (9)	0.0140 (8)	0.0186 (9)	-0.0003 (7)	0.0013 (7)	-0.0068 (7)
N2	0.0137 (8)	0.0124 (8)	0.0152 (9)	0.0003 (7)	-0.0017 (7)	-0.0059 (7)
N3	0.0198 (9)	0.0161 (9)	0.0148 (9)	-0.0019 (7)	0.0007 (8)	-0.0069 (7)

## supplementary materials

---

N4	0.0171 (9)	0.0202 (9)	0.0200 (10)	0.0007 (7)	0.0010 (8)	-0.0107 (8)
N5	0.0177 (10)	0.0272 (10)	0.0232 (10)	0.0003 (8)	0.0035 (8)	-0.0065 (8)
N6	0.0252 (11)	0.0283 (10)	0.0178 (10)	-0.0040 (8)	0.0061 (9)	-0.0034 (8)
C1	0.0242 (12)	0.0159 (10)	0.0304 (13)	-0.0034 (9)	-0.0040 (10)	-0.0091 (10)
C2	0.0256 (12)	0.0136 (10)	0.0220 (12)	-0.0013 (9)	0.0013 (10)	-0.0082 (9)
C3	0.0154 (11)	0.0178 (10)	0.0224 (12)	0.0005 (8)	0.0033 (9)	-0.0096 (9)
C4	0.0172 (11)	0.0154 (10)	0.0244 (12)	0.0043 (9)	-0.0036 (10)	-0.0057 (9)
C5	0.0198 (11)	0.0189 (10)	0.0141 (11)	0.0002 (9)	-0.0040 (9)	-0.0040 (9)
C6	0.0170 (11)	0.0184 (10)	0.0166 (11)	-0.0009 (8)	-0.0001 (9)	-0.0094 (9)
C7	0.0133 (10)	0.0127 (9)	0.0185 (11)	-0.0009 (8)	-0.0019 (9)	-0.0081 (8)
C8	0.0125 (10)	0.0120 (9)	0.0164 (10)	-0.0012 (8)	-0.0001 (8)	-0.0049 (8)
C9	0.0155 (10)	0.0155 (10)	0.0148 (10)	-0.0016 (8)	0.0002 (9)	-0.0064 (8)
C10	0.0133 (10)	0.0142 (9)	0.0177 (11)	-0.0030 (8)	0.0028 (9)	-0.0076 (8)
C11	0.0141 (10)	0.0130 (9)	0.0174 (11)	0.0004 (8)	0.0003 (9)	-0.0041 (8)
C12	0.0124 (10)	0.0153 (10)	0.0151 (10)	-0.0014 (8)	0.0003 (8)	-0.0062 (8)
C13	0.0182 (11)	0.0111 (9)	0.0171 (11)	-0.0009 (8)	0.0005 (9)	-0.0046 (8)
C14	0.0173 (11)	0.0178 (10)	0.0199 (11)	0.0039 (9)	-0.0038 (9)	-0.0078 (9)
C15	0.0208 (11)	0.0162 (10)	0.0226 (12)	0.0007 (9)	-0.0102 (10)	-0.0049 (9)
C16	0.0287 (12)	0.0208 (11)	0.0155 (11)	-0.0032 (10)	-0.0044 (10)	-0.0075 (9)
C17	0.0230 (12)	0.0212 (11)	0.0181 (11)	-0.0026 (9)	0.0011 (10)	-0.0095 (9)
C18	0.0120 (10)	0.0163 (10)	0.0186 (11)	-0.0013 (8)	0.0001 (9)	-0.0097 (9)
C19	0.0156 (10)	0.0154 (10)	0.0179 (11)	0.0001 (8)	-0.0007 (9)	-0.0063 (8)
C20	0.0160 (11)	0.0203 (10)	0.0147 (11)	-0.0020 (9)	0.0011 (9)	-0.0063 (9)
C21	0.0153 (10)	0.0146 (10)	0.0224 (12)	0.0014 (8)	-0.0006 (9)	-0.0075 (9)
C22	0.0133 (10)	0.0172 (10)	0.0154 (10)	-0.0006 (8)	0.0007 (9)	-0.0057 (8)
C23	0.0216 (12)	0.0164 (10)	0.0131 (10)	-0.0041 (9)	0.0058 (9)	-0.0050 (8)
C24	0.0131 (10)	0.0261 (11)	0.0228 (12)	0.0001 (9)	-0.0008 (9)	-0.0129 (10)

### *Geometric parameters (Å, °)*

Sn1—C2	2.102 (2)	C5—C6	1.382 (3)
Sn1—C1	2.110 (2)	C5—H5	0.9500
Sn1—N6	2.2258 (18)	C6—C7	1.386 (3)
Sn1—N5	2.2645 (19)	C6—H6	0.9500
Sn1—N3	2.5246 (17)	C7—C8	1.481 (3)
Sn1—N1	2.5418 (17)	C8—C9	1.385 (3)
Sn1—N2	2.5630 (17)	C9—C10	1.395 (3)
S1—C23	1.632 (2)	C9—H9	0.9500
S2—C24	1.624 (2)	C10—C11	1.393 (3)
N1—C3	1.343 (3)	C10—C18	1.478 (3)
N1—C7	1.351 (3)	C11—C12	1.387 (3)
N2—C8	1.342 (3)	C11—H11	0.9500
N2—C12	1.345 (3)	C12—C13	1.483 (3)
N3—C17	1.340 (3)	C13—C14	1.389 (3)
N3—C13	1.349 (3)	C14—C15	1.391 (3)
N4—C21	1.332 (3)	C14—H14	0.9500
N4—C20	1.337 (3)	C15—C16	1.372 (3)
N5—C23	1.161 (3)	C15—H15	0.9500
N6—C24	1.164 (3)	C16—C17	1.388 (3)

C1—H1A	0.9800	C16—H16	0.9500
C1—H1B	0.9800	C17—H17	0.9500
C1—H1C	0.9800	C18—C22	1.389 (3)
C2—H2A	0.9800	C18—C19	1.393 (3)
C2—H2B	0.9800	C19—C20	1.385 (3)
C2—H2C	0.9800	C19—H19	0.9500
C3—C4	1.383 (3)	C20—H20	0.9500
C3—H3	0.9500	C21—C22	1.382 (3)
C4—C5	1.383 (3)	C21—H21	0.9500
C4—H4	0.9500	C22—H22	0.9500
C2—Sn1—C1	173.66 (8)	C7—C6—C5	119.20 (19)
C2—Sn1—N6	95.00 (8)	C7—C6—H6	120.4
C1—Sn1—N6	88.97 (8)	C5—C6—H6	120.4
C2—Sn1—N5	94.55 (8)	N1—C7—C6	122.26 (18)
C1—Sn1—N5	90.97 (8)	N1—C7—C8	116.12 (18)
N6—Sn1—N5	80.75 (7)	C6—C7—C8	121.57 (18)
C2—Sn1—N3	84.97 (7)	N2—C8—C9	122.46 (18)
C1—Sn1—N3	91.04 (7)	N2—C8—C7	115.30 (17)
N6—Sn1—N3	77.94 (6)	C9—C8—C7	122.25 (18)
N5—Sn1—N3	158.55 (6)	C8—C9—C10	119.37 (19)
C2—Sn1—N1	85.50 (7)	C8—C9—H9	120.3
C1—Sn1—N1	92.67 (7)	C10—C9—H9	120.3
N6—Sn1—N1	158.04 (6)	C11—C10—C9	117.91 (18)
N5—Sn1—N1	77.33 (6)	C11—C10—C18	120.97 (18)
N3—Sn1—N1	123.89 (5)	C9—C10—C18	121.12 (19)
C2—Sn1—N2	97.06 (7)	C12—C11—C10	119.39 (19)
C1—Sn1—N2	76.72 (7)	C12—C11—H11	120.3
N6—Sn1—N2	138.11 (6)	C10—C11—H11	120.3
N5—Sn1—N2	137.58 (6)	N2—C12—C11	122.32 (19)
N3—Sn1—N2	63.44 (5)	N2—C12—C13	114.86 (17)
N1—Sn1—N2	63.18 (5)	C11—C12—C13	122.81 (18)
C3—N1—C7	117.48 (17)	N3—C13—C14	122.62 (19)
C3—N1—Sn1	122.53 (13)	N3—C13—C12	115.37 (18)
C7—N1—Sn1	119.96 (12)	C14—C13—C12	121.93 (19)
C8—N2—C12	118.48 (17)	C15—C14—C13	118.8 (2)
C8—N2—Sn1	117.17 (13)	C15—C14—H14	120.6
C12—N2—Sn1	116.78 (13)	C13—C14—H14	120.6
C17—N3—C13	117.84 (18)	C16—C15—C14	118.5 (2)
C17—N3—Sn1	120.77 (14)	C16—C15—H15	120.7
C13—N3—Sn1	121.07 (13)	C14—C15—H15	120.7
C21—N4—C20	116.62 (17)	C15—C16—C17	119.7 (2)
C23—N5—Sn1	176.01 (17)	C15—C16—H16	120.2
C24—N6—Sn1	159.06 (18)	C17—C16—H16	120.2
Sn1—C1—H1A	109.5	N3—C17—C16	122.5 (2)
Sn1—C1—H1B	109.5	N3—C17—H17	118.7
H1A—C1—H1B	109.5	C16—C17—H17	118.7
Sn1—C1—H1C	109.5	C22—C18—C19	117.38 (18)
H1A—C1—H1C	109.5	C22—C18—C10	120.88 (18)
H1B—C1—H1C	109.5	C19—C18—C10	121.73 (18)

## supplementary materials

---

Sn1—C2—H2A	109.5	C20—C19—C18	119.26 (19)
Sn1—C2—H2B	109.5	C20—C19—H19	120.4
H2A—C2—H2B	109.5	C18—C19—H19	120.4
Sn1—C2—H2C	109.5	N4—C20—C19	123.52 (19)
H2A—C2—H2C	109.5	N4—C20—H20	118.2
H2B—C2—H2C	109.5	C19—C20—H20	118.2
N1—C3—C4	123.57 (19)	N4—C21—C22	124.19 (19)
N1—C3—H3	118.2	N4—C21—H21	117.9
C4—C3—H3	118.2	C22—C21—H21	117.9
C3—C4—C5	118.28 (19)	C21—C22—C18	119.01 (19)
C3—C4—H4	120.9	C21—C22—H22	120.5
C5—C4—H4	120.9	C18—C22—H22	120.5
C6—C5—C4	119.17 (19)	N5—C23—S1	178.9 (2)
C6—C5—H5	120.4	N6—C24—S2	178.9 (2)
C4—C5—H5	120.4		
C2—Sn1—N1—C3	-65.95 (16)	Sn1—N1—C7—C8	-6.5 (2)
C1—Sn1—N1—C3	120.14 (17)	C5—C6—C7—N1	1.4 (3)
N6—Sn1—N1—C3	26.3 (3)	C5—C6—C7—C8	-176.24 (19)
N5—Sn1—N1—C3	29.74 (16)	C12—N2—C8—C9	1.2 (3)
N3—Sn1—N1—C3	-146.81 (15)	Sn1—N2—C8—C9	-147.48 (16)
N2—Sn1—N1—C3	-166.20 (17)	C12—N2—C8—C7	-179.06 (17)
C2—Sn1—N1—C7	116.06 (16)	Sn1—N2—C8—C7	32.2 (2)
C1—Sn1—N1—C7	-57.86 (16)	N1—C7—C8—N2	-17.1 (3)
N6—Sn1—N1—C7	-151.74 (18)	C6—C7—C8—N2	160.63 (18)
N5—Sn1—N1—C7	-148.25 (16)	N1—C7—C8—C9	162.59 (18)
N3—Sn1—N1—C7	35.19 (16)	C6—C7—C8—C9	-19.7 (3)
N2—Sn1—N1—C7	15.81 (13)	N2—C8—C9—C10	1.2 (3)
C2—Sn1—N2—C8	-106.42 (14)	C7—C8—C9—C10	-178.43 (18)
C1—Sn1—N2—C8	74.85 (14)	C8—C9—C10—C11	-2.5 (3)
N6—Sn1—N2—C8	147.95 (14)	C8—C9—C10—C18	177.98 (18)
N5—Sn1—N2—C8	-1.71 (18)	C9—C10—C11—C12	1.3 (3)
N3—Sn1—N2—C8	172.83 (15)	C18—C10—C11—C12	-179.16 (18)
N1—Sn1—N2—C8	-25.11 (13)	C8—N2—C12—C11	-2.5 (3)
C2—Sn1—N2—C12	104.32 (15)	Sn1—N2—C12—C11	146.34 (16)
C1—Sn1—N2—C12	-74.41 (15)	C8—N2—C12—C13	176.96 (17)
N6—Sn1—N2—C12	-1.31 (19)	Sn1—N2—C12—C13	-34.2 (2)
N5—Sn1—N2—C12	-150.97 (14)	C10—C11—C12—N2	1.2 (3)
N3—Sn1—N2—C12	23.57 (13)	C10—C11—C12—C13	-178.18 (18)
N1—Sn1—N2—C12	-174.37 (16)	C17—N3—C13—C14	1.2 (3)
C2—Sn1—N3—C17	62.31 (16)	Sn1—N3—C13—C14	174.64 (15)
C1—Sn1—N3—C17	-122.63 (16)	C17—N3—C13—C12	-175.43 (18)
N6—Sn1—N3—C17	-33.91 (16)	Sn1—N3—C13—C12	-2.0 (2)
N5—Sn1—N3—C17	-27.3 (3)	N2—C12—C13—N3	24.1 (3)
N1—Sn1—N3—C17	143.45 (14)	C11—C12—C13—N3	-156.48 (19)
N2—Sn1—N3—C17	162.79 (17)	N2—C12—C13—C14	-152.59 (19)
C2—Sn1—N3—C13	-110.95 (16)	C11—C12—C13—C14	26.9 (3)
C1—Sn1—N3—C13	64.12 (16)	N3—C13—C14—C15	-2.0 (3)
N6—Sn1—N3—C13	152.84 (16)	C12—C13—C14—C15	174.45 (19)
N5—Sn1—N3—C13	159.42 (17)	C13—C14—C15—C16	1.4 (3)



N1—Sn1—N3—C13	-29.80 (17)	C14—C15—C16—C17	-0.2 (3)
N2—Sn1—N3—C13	-10.47 (14)	C13—N3—C17—C16	0.1 (3)
C2—Sn1—N6—C24	-150.2 (5)	Sn1—N3—C17—C16	-173.39 (15)
C1—Sn1—N6—C24	24.8 (5)	C15—C16—C17—N3	-0.5 (3)
N5—Sn1—N6—C24	116.0 (5)	C11—C10—C18—C22	35.8 (3)
N3—Sn1—N6—C24	-66.5 (5)	C9—C10—C18—C22	-144.7 (2)
N1—Sn1—N6—C24	119.4 (5)	C11—C10—C18—C19	-143.3 (2)
N2—Sn1—N6—C24	-43.8 (5)	C9—C10—C18—C19	36.3 (3)
C7—N1—C3—C4	1.3 (3)	C22—C18—C19—C20	-1.2 (3)
Sn1—N1—C3—C4	-176.70 (15)	C10—C18—C19—C20	177.93 (19)
N1—C3—C4—C5	0.7 (3)	C21—N4—C20—C19	-0.7 (3)
C3—C4—C5—C6	-1.7 (3)	C18—C19—C20—N4	1.5 (3)
C4—C5—C6—C7	0.7 (3)	C20—N4—C21—C22	-0.4 (3)
C3—N1—C7—C6	-2.4 (3)	N4—C21—C22—C18	0.7 (3)
Sn1—N1—C7—C6	175.74 (15)	C19—C18—C22—C21	0.1 (3)
C3—N1—C7—C8	175.36 (18)	C10—C18—C22—C21	-178.98 (19)

Fig. 1

